

# THE CALCULATION OF OVERLAP INTEGRALS BETWEEN SLATER-TYPE ORBITALS WITH NEARLY EQUAL EXPONENTS

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It is shown that the usual formulas for the auxiliary functions  $B_n(\alpha)$ , necessary for the calculation of overlap integrals, give inaccurate results for small values of  $\alpha$ . Attention is drawn to better methods.

In calculating overlap integrals between Slater-type orbitals, e.g. from the master formulas of Mulliken et al. [1], one needs values for the auxiliary functions  $B_n(\alpha)$  defined by

$$B_n(\alpha) = \int_{-1}^1 x^n e^{-\alpha x} dx. \quad (1)$$

The  $B_n$ 's are given explicitly by

$$B_n(\alpha) = \frac{n!}{\alpha^{n+1}} \left[ e^\alpha \sum_{k=0}^n \frac{(-\alpha)^k}{k!} - e^{-\alpha} \sum_{k=0}^n \frac{\alpha^k}{k!} \right]. \quad (2)$$

But

$$B_n(0) = \begin{cases} \frac{2}{n+1} & \text{if } n \text{ is even;} \\ 0 & \text{if } n \text{ is odd.} \end{cases} \quad (3)$$

Therefore, Mulliken et al. give besides the general formula for each overlap integral another one for the case  $\alpha = 0$ . Alternatively a recursion formula can be used:

$$B_n(\alpha) = \frac{1}{\alpha} [n B_{n-1}(\alpha) - e^{-\alpha} + (-1)^n e^\alpha]. \quad (4)$$

Apart from a few exceptions [2-4] it is often not realised that both eq. (2) and (4) give inaccurate results for small values of  $\alpha$ . In such a case the use of eq. (2) involves the subtraction of two large quantities with a resulting loss in significant figures. The same problem is implicitly present in the recursion formula (4). The actual situation depends on the word length of the computer, the value of  $n$  and the smallness of  $\alpha$ . For second-row elements values of  $\alpha < 0.1$  can easily arise when slightly different exponents are used for 2s- and 2p-orbitals. For example, in calculating the 2s-2p $\sigma$  overlap between two double-bonded carbon atoms using the best atomic single exponents of Clementi and Raimondi [5], we have  $\alpha \approx 0.05$ . Some values of  $B_n(0.05)$  are given in table 1. With increasing  $n$  important deviations from the accurate values occur, and for  $n > 4$  not even the leading figure is correct. Although this should be evident to anybody con-

Table 1  
Some values of  $B_n(0.05)$  computed with 11 figures and rounded to 7. Parentheses contain powers of 10

	eq. (2)	eq. (4)	accurate
$B_0$	2. 000 833 (0)	2. 000 833 (0)	2. 000 833 (0)
$-B_1$	3. 334 166 (-2)	3. 334 166 (-2)	3. 334 167 (-2)
$B_2$	6. 671 669 (-1)	6. 671 670 (-1)	6. 671 667 (-1)
$-B_3$	1. 999 998 (-2)	1. 999 274 (-2)	2. 000 595 (-2)
$B_4$	4. 019 470 (-1)	4. 014 146 (-1)	4. 003 572 (-1)
$-B_5$	7. 812 500 (-3)	8. 982 866 (-2)	1. 429 034 (-2)
$B_6$	4. 000 000 (0)	1. 297 433 (1)	2. 859 921 (-1)

cerned with molecular computations, many programmes from the Quantum Chemistry Programme Exchange show that this is not generally true.

Corbató [2] has given a complicated procedure in which he uses the relation of the  $B_n$ 's to the spherical Bessel functions in order to circumvent these problems. However, a much simpler solution to the problem was given by Zener and Guillemin [6] as early as 1929. They expanded the exponential function in eq. (1) directly to give

even  $n$ :

$$B_n(\alpha) = 2 \sum_{k=0}^{\infty} \frac{\alpha^k}{k!(k+n+1)} \quad (k = \text{even}) ; \quad (5)$$

odd  $n$ :

$$B_n(\alpha) = -2 \sum_{k=1}^{\infty} \frac{\alpha^k}{k!(k+n+1)} \quad (k = \text{odd}) . \quad (6)$$

Using these expressions terms of the same sign are added together so that the  $B_n$ 's can be calculated to any desired accuracy. Furthermore eqs. (5) and (6) are equally valid for  $\alpha = 0$ , and there is no need for a special formula for the overlap integral to cover this case. Therefore the easiest correct way to evaluate the  $B_n$ 's is probably to use the series (5) and (6) for  $\alpha < 1$ , in which region there is a rapid convergence, and the re-

ursion formula (4) for  $\alpha > 1$ . The choice of  $\alpha = 1$  is arbitrary, of course. In our computations with 11 figures eqs. (2) or (4) still give 8 significant figures for  $B_4(1)$ , but only 3 for  $B_{10}(1)$ . Thus the given procedure is acceptable for second-row atoms where  $n \leq 4$ . If  $B_n$ 's for higher values of  $n$  are required it is necessary to extend the use of eqs. (5) and (6) to values of  $\alpha > 1$ , or alternatively the loss of significant figures in applying eq. (4) can be circumvented by using the recursion formula downwards (decreasing  $n$ ). In that case eqs. (5) or (6) can be used to calculate the starting value. Used in this way the recursion formula applies to the case  $\alpha = 0$  as well.

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